§172.665 Gellan gum.

The food additive gellan gum may be safely used in food in accordance with the following prescribed conditions:

- (a) The additive is a high molecular weight polysaccharide gum produced from *Pseudomonas elodea* by a pure culture fermentation process and purified by recovery with isopropyl alcohol. It is composed of tetrasaccharide repeat units, each containing one molecule of rhamnose and glucuronic acid, and two molecules of glucose. The glucuronic acid is neutralized to a mixed potassium, sodium, calcium, and magnesium salt. The polysaccharide may contain acyl (glyceryl and acetyl) groups as the O-glycosidically linked esters.
- (b) The strain of *P. elodea* is non-pathogenic and nontoxic in man and animals.
- (c) The additive is produced by a process that renders it free of viable cells of *P. elodea*.
- (d) The additive meets the following specifications:
- (1) Positive for gellan gum when subjected to the following identification tests:
- (i) A 1-percent solution is made by hydrating 1 gram of gellan gum in 99 milliliters of distilled water. The mixture is stirred for about 2 hours, using a motorized stirrer and a propeller-type stirring blade. A small amount of the above solution is drawn into a wide bore pipet and transferred into a solution of 10-percent calcium chloride. A tough worm-like gel will form instantly.
- (ii) To the 1-percent distilled water solution prepared for identification test (i), 0.50 gram of sodium chloride is added. The solution is heated to 80 °C with stirring, held at 80 °C for 1 minute, and allowed to cool to room temperature without stirring. A firm gel will form.
- (2) Residual isopropyl alcohol (IPA) not to exceed 0.075 percent as determined by the procedure described in the Xanthan Gum monograph, the "Food Chemicals Codex," 3d Ed. (1981), p. 347, which reads:

IPA Standard Solution. Transfer 500.0 milligrams (mg) chromatographic quality isopropyl alcohol into a 50-milliliter (ml) volumetric flask, dilute to volume with water, and mix. Pipet 10 ml of this solution into a

100-ml volumetric flask, dilute to volume with water, and mix.

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Tert-Butyl Alcohol (TBA) Standard Solution. Transfer 500.0 mg of chromatographic quality tert-butyl alcohol into a 50-ml volumetric flask, dilute to volume with water, and mix. Pipet 10 ml of this solution into a 100-ml volumetric flask, dilute to volume with water, and mix.

Mixed Standard Solution. Pipet 4 ml each of the IPA Standard Solution and of the TBA Standard Solution into a 125-ml graduated Erlenmeyer flask, dilute to about 100 ml with water, and mix. This solution contains approximately 40 micrograms (µg) each of isopropyl alcohol and tert-butyl alcohol per ml.

Sample Preparation. Disperse 1 ml of a suitable antifoam emulsion, such as Dow-Corning G-10 or equivalent, in 200 ml of water contained in a 1000-ml 24/40 round-bottom distilling flask. Add about 5 grams (g) of the sample, accurately weighed. and shake for 1 hour on a wrist-action mechanical shaker. Connect the flask to a fractionating column and distill about 100 ml, adjusting the heat so that foam does not enter the column. Add 4.0 ml of TBA Standard Solution to the distillate to obtain the Sample Preparation.

Procedure. Inject about 5 microliters (µl) of the Mixed Standard Solution into a suitable gas chromatograph equipped with a flame-ionization detector and a 1.8 meters (m)×3.2 millimeters (mm) stainless steel column packed with 80/100-mesh Porapak QS or equivalent. The carrier is helium flowing at 80 ml per minute (min). The injection port temperature is 200 °C, the column temperature is 200 °C, and the detector temperature is 200 °C. The retention time of isopropyl alcohol is about 2 min, and that of tert-butyl alcohol about 3 min.

Determine the areas of IPA and TBA peaks, and calculate the response factor, f, by the formula $A_{\rm IPA}/A_{\rm TBA}$, in which $A_{\rm IPA}$ is the area of the isopropyl alcohol peak, and $A_{\rm TBA}$ is the area of the *tert*-butyl alcohol peak.

Similarly, inject about 5 μ l of the Sample Preparation, and determine the peak areas, recording the area of the isopropyl alcohol peak as a $a_{\rm IPA}$, and that of the tert-butyl alcohol peak as a $a_{\rm TBA}$. Calculate the isopropyl alcohol content, in parts per million (ppm), in the sample taken by the formula ($a_{\rm IPA} \times 4000$)/ ($fxa_{\rm TBA} \times W$), in which W is the weight of the sample taken in grams.

- (e) The additive is used or intended for use in accordance with current good manufacturing practice as a stabilizer and thickener as defined in §170.3(o) (28) of this chapter. The additive may be used in foods where standards of identity established under section 401 of the Federal Food, Drug, and Cosmetic Act do not preclude such use.
 - (f) To assure safe use of the additive:

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- (1) The label of its container shall bear, in addition to other information required by the Federal Food, Drug, and Cosmetic Act, the name of the additive and the designation "food grade".
- (2) The label or labeling of the food additive container shall bear adequate directions for use.

[55 FR 39614, Sept. 28, 1990, as amended at 57 FR 55445, Nov. 25, 1992]

§172.695 Xanthan gum.

The food additive xanthan gum may be safely used in food in accordance with the following prescribed conditions:

- (a) The additive is a polysaccharide gum derived from *Xanthomonas campestris* by a pure-culture fermentation process and purified by recovery with isopropyl alcohol. It contains Deglucose, D-mannose, and D-glucuronic acid as the dominant hexose units and is manufactured as the sodium, potassium, or calcium salt.
- (b) The strain of *Xanthomonas* campestris is nonpathogenic and nontoxic in man or other animals.
- (c) The additive is produced by a process that renders it free of viable cells of *Xanthomonas campestris*.
- (d) The additive meets the following specifications:
- (1) Residual isopropyl alcohol not to exceed 750 parts per million.
- (2) An aqueous solution containing 1 percent of the additive and 1 percent of potassium chloride stirred for 2 hours has a minimum viscosity of 600 centipoises at 75 °F, as determined by Brookfield Viscometer, Model LVF (or equivalent), using a No. 3 spindle at 60 °F and 150 °F is in the range of 1.02 to 1.45.
- (3) Positive for xanthan gum when subjected to the following procedure:

LOCUST BEAN GUM GEL TEST

Blend on a weighing paper or in a weighing pan 1.0 gram of powdered locust bean gum with 1.0 gram of the powdered polysaccharide to be tested. Add the blend slowly (approximately $\frac{1}{2}$ minute) at the point of maximum agitation to a stirred solution of 200 milliliters of distilled water previously heated to 80 °C in a 400-milliliter beaker. Continue mechanical stirring until the mixture is in solution, but stir for a minimum time of 30 min-

utes. Do not allow the water temperature to drop below 60 $^{\circ}\text{C}.$

Set the beaker and its contents aside to cool in the absence of agitation. Allow a minimum time of 2 hours for cooling. Examine the cooled beaker contents for a firm rubbery gel formation after the temperature drops below $40~^{\circ}\text{C}$.

In the event that a gel is obtained, make up a 1 percent solution of the polysaccharide to be tested in 200 milliliters of distilled water previously heated to 80 °C (omit the locust bean gum). Allow the solution to cool without agitation as before. Formation of a gel on cooling indicates that the sample is a gelling polysaccharide and not xanthan gum.

Record the sample as "positive" for xanthan gum if a firm, rubbery gel forms in the presence of locust bean gum but not in its absence. Record the sample as "negative" for xanthan gum if no gel forms or if a soft or brittle gel forms both with locust bean gum and in a 1 percent solution of the sample (containing no locust bean gum).

(4) Positive for xanthan gum when subjected to the following procedure:

PYRUVIC ACID TEST

Pipet 10 milliliters of an 0.6 percent solution of the polysaccharide in distilled water (60 milligrams of water-soluble gum) into a 50-milliliter flask equipped with a standard taper glass joint. Pipet in 20 milliliters of 1N hydrochloric acid. Weigh the flask. Reflux the mixture for 3 hours. Take precautions to avoid loss of vapor during the refluxing. Cool the solution to room temperature. Add distilled water to make up any weight loss from the flask contents.

Pipet 1 milliliter of a 2,4-dinitrophenylhydrazine reagent (0.5 percent in 2N hydrochloric acid) into a 30-milliliter separatory funnel followed by a 2-milliliter aliquot (4 milligrams of water-soluble gum) of the polysaccharide hydrolyzate. Mix and allow the reaction mixture to stand at room temperature for 5 minutes. Extract the mixture with 5 milliliters of ethyl acetate. Discard the aqueous layer.

Extract the hydrazone from the ethyl acetate with three 5 milliliter portions of 10 percent sodium carbonate solution. Dilute the combined sodium carbonate extracts to 100 milliliters with additional 10 percent sodium carbonate in a 10-milliliter volumetric flask. Measure the optical density of the sodium carbonate solution at 375 millimicrons.

Compare the results with a curve of the optical density versus concentration of an authentic sample of pyruvic acid that has been run through the procedure starting with the preparation of the hydrazone.

Record the percent by weight of pyruvic acid in the test polysaccharide. Note "positive" for xanthan gum if the sample contains